Div. App. of 09/985,642

Preliminary Amdt. dated Nov. , 2003

## **Amendments to the Claims:**

This listing of claims will replace all prior versions, and listings of claims in the application:

## **Listing of Claims:**

Please cancel Claims 1 to 20, without prejudice, and add the following claims:

- 1.-20. (Deleted)
- 21. (New) Nanocomposite materials photoluminescent at ambient temperature produced by a sol-gel process comprising:
- preparing an aqueous or hydroalcoholic mixture containing a silicon alkoxide, an additional component A, and an acidic catalyst, wherein the molar ratio between water molecules and silicon atoms is equal to or higher than 4;
  - causing the mixture to gel thereby obtaining a wet gel;
  - causing said wet gel to dry; and
- densifing the thus obtained dry gel by means of a thermal treatment having a maximum temperature from 1200°C to 1400°C; where
- the additional component A is a dialkyldialkoxysilane,  $R_2$ -Si-(OR')<sub>2</sub>, or an alkyltrialkoxysilane, R-Si-(OR')<sub>3</sub>, wherein R and R' radicals are not aromatic; and
- in the range from 300°C to 800 °C the thermal treatment is carried out under an atmosphere made up of pure HCl or a mixture containing at least 5% by volume of HCl in an inert gas, said atmosphere being anhydrous and not containing oxygen.

Div. App. of 09/985,642

Preliminary Amdt. dated Nov. \_\_\_\_\_, 2003

- 22. (New) The nanocomposite materials according to Claim 21 wherein the silicon alkoxide is tetramethoxysilane or tetraethoxysilane.
- 23. (New) The nanocomposite materials according to Claim 21 wherein the acidic catalyst is HCl.
- 24. (New) The nanocomposite materials according to Claim 21 wherein the –R groups of the additional component A are selected from the group consisting of methyl, ethyl, propyl and butyl, and the –OR groups of the additional component A are selected from the group consisting of methoxy, ethoxy, propoxy and butoxy.
- 25. (New) The nanocomposite materials according to Claim 24 wherein the additional component A is selected from the group consisting of methyltrimethoxysilane and methyltriethoxysilane.
- 26. (New) The nanocomposite materials according to Claim 21 wherein the molar ratio between the silicon alkoxide and the additional component A is from 1.86 to 999.
- 27. (New) The nanocomposite materials according to Claim 26 wherein said molar ratio is from 2.33 to 9.
- 28. (New) The nanocomposite materials according to Claim 21 wherein pyrogenic silica is present.

Div. App. of 09/985,642

Preliminary Amdt. dated Nov. , 2003

- 29. (New) The nanocomposite materials according to Claim 21 wherein gelation is obtained by raising the pH of the mixture.
- 30. (New) The nanocomposite materials according to Claim 29 wherein raising the pH of the mixture is realized by adding a solution of ammonia.
- 31. (New) The nanocomposite materials according to Claim 21 wherein sol gelation is obtained by raising the temperature to a value in the range of 40°C to 60°C.
- 32. (New) The nanocomposite materials according to Claim 21 wherein drying of the wet gel is obtained by evaporation of liquid in pores of the gel.
- 33. (New) The nanocomposite materials according to Claim 21 wherein drying of the wet gel is obtained by supercritical extraction of liquid in pores of the gel.
- 34. (New) The nanocomposite materials according to Claim 23 wherein, before the supercritical extraction, the wet gel is subjected to an operation of exchange of liquid in pores of the gel.
- 35. (New) The nanocomposite materials according to Claim 21 wherein the sol is deposited in form of a thin layer on a substrate by immersing the substrate in the sol and then extracting the substrate from the sol.

Div. App. of 09/985,642

Preliminary Amdt. dated Nov. \_\_\_\_\_, 2003

- 36. (New) The nanocomposite materials according to Claim 21 wherein the sol is deposited in form of a thin layer on a substrate by depositing a drop of the sol on the substrate and rotating the substrate at high speed.
- 37. (New) Supported thin layers of nanocomposite materials photoluminescent at ambient temperature produced by
- preparing an aqueous or hydroalcoholic mixture containing a silicon alkoxide, an additional component A, and an acidic catalyst, wherein the molar ratio between water molecules and silicon atoms is equal to or higher than 4;
  - causing the mixture to gel thereby obtaining a wet gel;
  - causing said wet gel to dry; and
- densifing the thus obtained dry gel by means of a thermal treatment having a maximum temperature from 1200°C to 1400°C; where
- the additional component A is a dialkyldialkoxysilane,  $R_2$ -Si-(OR')<sub>2</sub>, or an alkyltrialkoxysilane, R-Si-(OR')<sub>3</sub>, wherein R and R' radicals are not aromatic; and
- in the range from 300°C to 800 °C the thermal treatment is carried out under an atmosphere made up of pure HCl or a mixture containing at least 5% by volume of HCl in an inert gas, said atmosphere being anhydrous and not containing oxygen,

wherein the sol is deposited in form of a thin layer on a substrate by immersing the substrate in the sol and then extracting the substrate from the sol.

Div. App. of 09/985,642

Preliminary Amdt. dated Nov. \_\_\_\_\_, 2003

- 38. (New) Supported thin layers of nanocomposite materials photoluminescent at ambient temperature produced by
- preparing an aqueous or hydroalcoholic mixture containing a silicon alkoxide, an additional component A, and an acidic catalyst, wherein the molar ratio between water molecules and silicon atoms is equal to or higher than 4;
  - causing the mixture to gel thereby obtaining a wet gel;
  - causing said wet gel to dry; and
- densifing the thus obtained dry gel by means of a thermal treatment having a maximum temperature from 1200°C to 1400°C; where
- the additional component A is a dialkyldialkoxysilane,  $R_2$ -Si-(OR')<sub>2</sub>, or an alkyltrialkoxysilane, R-Si-(OR')<sub>3</sub>, wherein R and R' radicals are not aromatic; and
- in the range from 300°C to 800°C the thermal treatment is carried out under an atmosphere made up of pure HCl or a mixture containing at least 5% by volume of HCl in an inert gas, said atmosphere being anhydrous and not containing oxygen,

wherein the sol is deposited in form of a thin layer on a substrate by depositing a drop of the sol on the substrate and rotating the substrate at high speed.